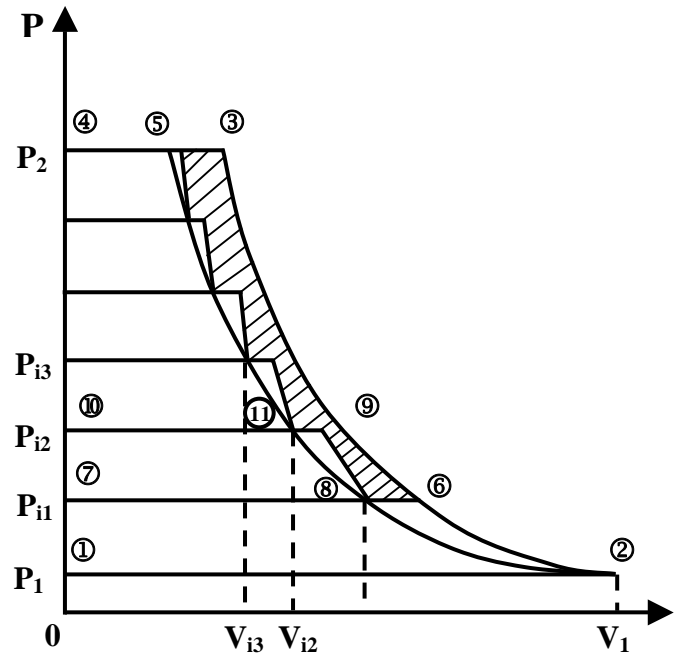


- ▶ A-line ①→② represents the suction stroke of the first stage where a volume ( $V_1$ ) of gas is admitted at a pressure ( $P_1$ ).
- ▶ B-line ②→⑥ represents an isentropic compression to a pressure ( $P_{i1}$ ).
- ▶ C-line ⑥→⑦ represents the delivery of the gas from the first stage at a constant pressure ( $P_{i1}$ ).
- ▶ D-line ⑦→⑧ represents the suction stroke of the second stage. The volume of the gas has the reduced in the inter-stage cooler to ( $V_{i1}$ ), that which would have been obtained as a result of an isothermal compression to ( $P_{i2}$ ).



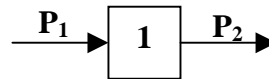
- ▶ E-line ⑧→⑨ represents an isentropic compression in the second stage from a pressure ( $P_{i1}$ ) to a pressure ( $P_{i2}$ ).
- ▶ F-line ⑨→⑩ represents the delivery stroke of the second stage.
- ▶ G-line ⑩→⑪ represents the suction stroke of the third stage point ⑪ again lyses on the line ②→⑤ that representing an isothermal compression.

It is seen that the overall work done on the gas is intermediate between that for a single stage isothermal compression and that for isentropic compression. The net saving in energy is shown as the shaded area in the last Figure.

### The Total Work Done for Multistage Compressors

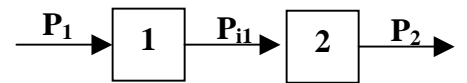
🔔 The total work done for compression the gas from  $P_1$  to  $P_2$  in an ideal single stage is,

$$W = \frac{\gamma}{\gamma-1} P_1 V_1 \left[ \left( \frac{P_2}{P_1} \right)^{(\gamma-1)/\gamma} - 1 \right]$$



🔔 The total work done for compression the gas from  $P_1$  to  $P_2$  in an ideal two stages is,

$$W = \frac{\gamma}{\gamma-1} P_1 V_1 \left[ \left( \frac{P_{i1}}{P_1} \right)^{(\gamma-1)/\gamma} - 1 \right] + \frac{\gamma}{\gamma-1} P_{i1} V_{i1} \left[ \left( \frac{P_2}{P_{i1}} \right)^{(\gamma-1)/\gamma} - 1 \right]$$

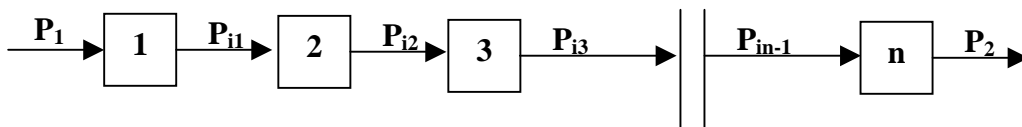


but for perfect inter-stage cooling i.e. at isothermal line  $P_1 V_1 = P_{i1} V_{i1} = \text{constant}$

$$\Rightarrow W = \frac{\gamma}{\gamma-1} P_1 V_1 \left[ \left\{ \left( \frac{P_{i1}}{P_1} \right)^{(\gamma-1)/\gamma} - 1 \right\} + \left\{ \left( \frac{P_2}{P_{i1}} \right)^{(\gamma-1)/\gamma} - 1 \right\} \right]$$

🔔 The total work done for compression the gas from  $P_1$  to  $P_2$  in an ideal n-stages is,

$$W = \frac{\gamma}{\gamma-1} P_1 V_1 \left[ \left( \frac{P_{i1}}{P_1} \right)^{(\gamma-1)/\gamma} - 1 \right] + \frac{\gamma}{\gamma-1} P_{i1} V_{i1} \left[ \left( \frac{P_{i2}}{P_{i1}} \right)^{(\gamma-1)/\gamma} - 1 \right] + \dots + \frac{\gamma}{\gamma-1} P_{in-1} V_{in-1} \left[ \left( \frac{P_2}{P_{in-1}} \right)^{(\gamma-1)/\gamma} - 1 \right]$$



for perfect inter-stage cooling  $P_1 V_1 = P_{i1} V_{i1} = P_{i2} V_{i2} = \dots = P_{in-1} V_{in-1} = \text{constant}$

$$W = \frac{\gamma}{\gamma-1} P_1 V_1 \left[ \left\{ \left( \frac{P_{i1}}{P_1} \right)^{(\gamma-1)/\gamma} - 1 \right\} + \left\{ \left( \frac{P_{i2}}{P_{i1}} \right)^{(\gamma-1)/\gamma} - 1 \right\} + \dots + \left\{ \left( \frac{P_2}{P_{in-1}} \right)^{(\gamma-1)/\gamma} - 1 \right\} \right]$$

$$\Rightarrow W = \frac{\gamma}{\gamma-1} P_1 V_1 \left[ \left( \frac{P_{i1}}{P_1} \right)^{(\gamma-1)/\gamma} + \left( \frac{P_{i2}}{P_{i1}} \right)^{(\gamma-1)/\gamma} + \dots + \left( \frac{P_2}{P_{in-1}} \right)^{(\gamma-1)/\gamma} - n \right]$$

The optimum values of intermediate pressures  $P_{i1}, P_{i2}, P_{i3}, \dots, P_{in-1}$  are so that **the compression ratio (r) is the same in each stage and equal work is then done in each stage.**

$$\text{i.e. } \frac{P_{i1}}{P_1} = \frac{P_{i2}}{P_{i1}} = \frac{P_{i3}}{P_{i2}} = \dots = \frac{P_2}{P_{in-1}} = r$$

$$\text{then } \left( \frac{P_2}{P_1} \right)^{\frac{1}{n}} = \frac{P_{i1}}{P_1} = \frac{P_{i2}}{P_{i1}} = \frac{P_{i3}}{P_{i2}} = \dots = \frac{P_2}{P_{in-1}} = r \text{ -----prove that}$$

$$\Rightarrow W = \frac{\gamma}{\gamma-1} P_1 V_1 \left[ n \left( \frac{P_2}{P_1} \right)^{(\gamma-1)/n\gamma} - n \right] \Rightarrow W = P_1 V_1 \frac{n\gamma}{\gamma-1} \left[ \left( \frac{P_2}{P_1} \right)^{(\gamma-1)/n\gamma} - 1 \right]$$

The effect of clearance volume can now be taken into account. If the clearance in the successive cylinders are  $C_1, C_2, C_3, \dots, C_n$  the theoretical volumetric efficiency of the first cylinder =  $[1 + C_1 - C_1 (P_2 / P_1)^{1/\gamma}]$ .

Assuming that the same compression ratio is used in each cylinder, then the theoretical volumetric efficiency of the first stage =  $[1 + C_1 - C_1 (P_2 / P_1)^{1/n\gamma}]$ .

If the swept volumes of the cylinders are  $V_{s1}, V_{s2}, V_{s3}, \dots$  the volume of the gas admitted to the first cylinder =  $V_{s1} [1 + C_1 - C_1 (P_2 / P_1)^{1/n\gamma}]$

The same mass of gas passes through each of the cylinders and, therefore, if the inter-stage coolers are assumed perfectly efficient, the ratio of the volumes of gas admitted to successive cylinder is  $(P_1 / P_2)^{1/n}$  [because lies on the isothermal line].

The volume of gas admitted to the second cylinder

$$= V_{s2} [1 + C_2 - C_2 (P_2 / P_1)^{1/n\gamma}] = V_{s1} [1 + C_1 - C_1 (P_2 / P_1)^{1/n\gamma}] (P_1 / P_2)^{1/n}$$

$$\Rightarrow \frac{V_{s1} [1 + C_2 - C_2 (P_2 / P_1)^{1/n\gamma}]}{V_{s2} [1 + C_1 - C_1 (P_2 / P_1)^{1/n\gamma}]} (P_2 / P_1)^{1/n}$$

In this manner the swept volume of each cylinder can be calculated in terms of  $V_{s1}$ , and  $C_1, C_2, \dots$ , and the cylinder dimensions determined.

$$\text{Let } V_1 = V_{s1} [1 + C_1 - C_1 (P_2 / P_1)^{1/n\gamma}], \quad V_2 = V_{s2} [1 + C_2 - C_2 (P_2 / P_1)^{1/n\gamma}]$$

Where,  $V_i$  : represents the volume of gas admitted to stage i.

But for perfectly cooled [i.e. isothermal]  $\Rightarrow P_1 V_1 = P_{i1} V_{i1} = P_{i2} V_{i2} = \dots = P_{in-1} V_{in-1}$

$$\Rightarrow P_1 V_{s1} [1 + C_1 - C_1 (P_2 / P_1)^{1/n\gamma}] = P_{i1} V_{s2} [1 + C_2 - C_2 (P_2 / P_1)^{1/n\gamma}]$$

$$\text{but } r = \frac{P_{i1}}{P_1} = \left( \frac{P_2}{P_1} \right)^{1/n}$$

$$\Rightarrow \frac{V_{s1}}{V_{s2}} = \frac{[1 + C_2 - C_2 (P_2 / P_1)^{1/n\gamma}]}{[1 + C_1 - C_1 (P_2 / P_1)^{1/n\gamma}]} \left( \frac{P_2}{P_1} \right)^{1/n}$$

### Example -8.12-

A single-acting air compressor supplies  $0.1 \text{ m}^3/\text{s}$  of air (at STP) compressed to 380 kPa from 101.3 kPa. If the suction temperature is 289 K, the stroke is 0.25 m, and the speed is 4 Hz, what is the cylinder diameter? Assume the cylinder clearance is 4% and compression and re-expansion are isentropic ( $\gamma=1.4$ ). What are the theoretical power requirements for the compression?

#### Solution:

Stroke (حركة من سلسلة حركات متسلسلة "متوالية ومتشابهة")

Volume of gas per stroke =  $(0.1 \text{ m}^3/\text{s})/4\text{s}^{-1} (289/273)$

$$= 0.0264 \text{ m}^3$$

$$= (V_1 - V_4) \equiv [\text{volume of gas admitted per cycle}]$$

$$P_2/P_1 = 380/101.3 = 3.75$$

$$(V_1 - V_4) = V_s [1 + C - C (P_2/P_1)^{1/\gamma}]$$

$$0.0264 = V_s [1 + 0.04 - 0.04(3.75)^{1/1.4}] \Rightarrow V_s = 0.0283 \text{ m}^3 = (V_1 - V_3) \equiv \text{volume of cylinder}$$

$$\text{Cross-section area of cylinder} = V_s/L_{\text{stroke}} = 0.0283/0.25 = 0.113 \text{ m}^2$$

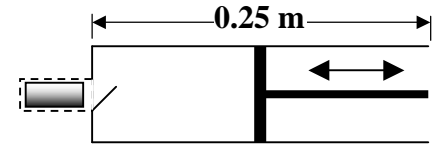
$$\Rightarrow \text{The diameter of cylinder} = [0.113/(\pi/4)]^{1/2} = 0.38 \text{ m}$$

$$W = \frac{\gamma}{\gamma-1} P_1 (V_1 - V_4) \left[ \left( \frac{P_2}{P_1} \right)^{(\gamma-1)/\gamma} - 1 \right]$$

for 1kg of gas that compressed [or per cycle]

$$\Rightarrow W = \frac{1.4}{0.4} (101.3 \times 10^3) (0.0264) [(3.75)^{0.4/1.4} - 1] = 4278 \text{ J / kg per stroke}$$

$$\text{The theoretical power required} = 4278 \text{ J/kg} (4\text{s}^{-1}) \text{ per stroke} = 17110 \text{ W} = 17.11 \text{ kW}$$



### Example -8.13-

Air at 290 K is compressed from 101.3 kPa to 2065 kPa in two-stage compressor operating with a mechanical efficiency of 85%. The relation between pressure and volume during the compression stroke and expansion of clearance gas is ( $PV^{1.25} = \text{constant}$ ). The compression ratio in each of the two cylinders is the same, and the inter-stage cooler may be assumed 100% efficient. If the clearance in the two cylinders are 4% and 5%, calculate:

- The work of compression per kg of air compressed;
- The isothermal efficiency;
- The isentropic efficiency;
- The ratio of swept volumes in the two cylinders.

#### Solution:

$$P_2/P_1 = 2065/101.3 = 20.4$$

$$V_1 = \frac{RT}{P_1 Mwt} = \frac{8314 (\text{Pa} \cdot \text{m}^3/\text{kmol} \cdot \text{K}) 290\text{K}}{(101.3 \times 10^3 \text{ Pa}) 29 \text{ kg/kmol}} = 0.82 \text{ (m}^3/\text{kg)}$$

For 100% efficient of cooler at inter-stage, the work of compression in multistage compressor of n-stages is;

$$W = P_1 V_1 \frac{n\gamma}{\gamma-1} \left[ \left( \frac{P_2}{P_1} \right)^{(\gamma-1)/n\gamma} - 1 \right] \Rightarrow W = (101.3 \times 10^3 \times 0.82) \frac{2(1.25)}{1.25-1} [(20.4)^{(0.25)/2.5} - 1] = 292.35 \frac{\text{kJ}}{\text{kg}}$$